

Worked Example

Use of the CO₂ ventilation model

Consider an office 27m × 21m × 3m(H) housing people at 9m²/person and a design ventilation rate of 10L/s/person. What is the CO₂ concentration at different hours of the day assuming that office hour is 9am to 6pm with lunch time at 1pm to 2pm. The office is purged with fresh air after work.

- c = concentration of carbon dioxide in the room (ppm)
 Q' = fresh air supply rate per occupant (m³/s) (10L/s/person = 0.01m³/s/person)
 V' = occupied volume in the room per occupant (m³)
 t = time after room is occupied (sec)
 c_a = concentration of carbon dioxide in outdoor ambient air (take 380ppm)
 \dot{V}_c = volume of carbon dioxide produced per occupant (take 4×10^{-6} m³/s)
 c_0 = concentration of carbon dioxide in the room at time $t = 0$ (ppm)

$$c = \left[\frac{10^6 \dot{V}_c}{Q'} + c_a \right] \left(1 - e^{-\frac{Q't}{V'}} \right) + c_0 e^{-\frac{Q't}{V'}}$$

$$\text{Volume of room} = 27\text{m} \times 21\text{m} \times 3\text{m} = 1701\text{m}^3$$

$$\text{The occupied volume per person } V' = 9\text{m}^2 \times 3\text{m} = 27\text{m}^3$$

Since the room is purged with fresh air overnight, therefore, it can be assumed that the CO₂ concentration at the start of the working at 9am = outdoor concentration = 380ppm

Consider after one hour (i.e. at 10am, $t = 3600\text{s}$)

Substituting into the equation:

$$\frac{Q't}{V'} = \frac{0.01 \times 3600}{27} = 1.33$$

$$e^{-\frac{Q't}{V'}} = e^{-1.33} = 0.264$$

$$c = \left[\frac{10^6 \times 4 \times 10^{-6}}{0.01} + 380 \right] (1 - 0.264) + 380 \times 0.264 = 674 \text{ ppm}$$

Using similar calculation, the CO₂ concentration at different hours before lunch is given by:

$$10\text{am} = 674\text{ppm}$$

$$11\text{am} = 752\text{ppm}$$

$$12\text{noon} = 773\text{ppm}$$

$$1\text{pm} = 778\text{ppm}$$

At lunch hour, assume no one inside the office, but ventilation remains at constant rate

$$c = \left[\frac{10^6 \times 0}{0.01} + 380 \right] (1 - 0.264) + 778 \times 0.264 = 485 \text{ ppm}$$

thus

at 2pm = 485ppm

The CO₂ concentration for afternoon office hour is done similarly except putting the initial CO₂ concentration at 485ppm instead of 380ppm in the morning.

The outdoor air change rate for the office

$$= \frac{0.01 \times 3600}{27} = 1.33 \text{ ACH}$$

Exercise

For energy saving purposes, the office in the worked example has reduced the ventilation rate such that a steady state 1000ppm CO₂ concentration can be achieved. What should be the ventilation rate (in L/s/person or ACH)?

Try also to derive the equations in steps.

Solution

Using

$$c = \left[\frac{10^6 \dot{V}_c}{Q'} + c_a \right] \left(1 - e^{-\frac{Q't}{V'}} \right) + c_0 e^{-\frac{Q't}{V'}} \quad \text{and inserting the necessary parameters}$$

$$1000 = \left[\frac{10^6 \times 4 \times 10^{-6}}{Q'} + 380 \right] (1 - e^{-n}) + 380 e^{-n} \quad \text{where } n = \frac{Q't}{V'}$$

For steady state, $e^{-n} \rightarrow e^{-\infty} \rightarrow 0$

$$1000 = \left[\frac{4}{Q'} + 380 \right]$$

Thus

$$Q' = 0.00645 \text{ m}^3/\text{s}/\text{person} = 6.5 \text{ L/s}/\text{person}$$

Air change rate

$$= \frac{0.00645 \times 3600}{27} = 0.86 \text{ ACH}$$